

Supplementary Information

Effect of temperature and solvents on thermo-physical properties of pyrimidine substituted thiazolidinone derivatives at three different temperatures

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Supplemental Materials

Content

- Synthetic procedure of AZ₁ and AZ₂
- Spectral analysis of AZ₁ and AZ₂
- IR, ¹H NMR, ¹³C NMR, and Mass spectra of compounds AZ₁ and AZ₂.

Procedure for synthesis of 5-(4-fluorobenzylidene)-2-(furan-2-yl)-3-(pyrimidin-2-yl)thiazolidin-4-one (AZ₁) and 2-(furan-2-yl)-5-(4-methoxybenzylidene)-3-(pyrimidin-2-yl)thiazolidin-4-one (AZ₂).

2-(furan-2-yl)-3-(pyrimidin-2-yl)thiazolidin-4-one was dissolved in ethanol (0.01 mol), and various aromatic aldehydes (0.01 mol) were then added. The catalyst for the reaction was sodium ethoxide. The reaction mixture was refluxed at 50–60 °C for 4–6 hours. The finished reaction mixture was poured onto crushed ice after being allowed to cool at ambient temperature to get the desired compounds (**AZ₁ and AZ₂**). The produced product underwent filtering, distilled water washing, vacuum oven drying, and recrystallization.

5-(4-fluorobenzylidene)-2-(furan-2-yl)-3-(pyrimidin-2-yl)thiazolidin-4-one, AZ₁: Yield 52%, solid, dirty white, m.p. 162-164°C. IR (KBr, ν_{\max} , cm⁻¹): 683 (C-S), 1177 (C-F), 1287 (C-O-C), 1334 (-C-N), 1598 (-C=N-), 1687 (C=O), 3066 (C-H). ¹H NMR spectrum (500 MHz, DMSO-d₆, δ ppm): 6.36 (t, J = 7.53 Hz, 1H, C=CH), 6.43 – 6.48 (m, 2H), 6.96 (t, J = 7.51 Hz, 1H, C=CH), 7.19 – 7.26 (m, 2H), 7.47 (d, J = 7.46 Hz, 1H, O-CH), 7.57 – 7.64

(m, 2H), 7.81 (s, 1H, C=CH), 8.42 (d, J = 7.53 Hz, 2H, CH=C-CH). ¹³C NMR spectrum (125 MHz, DMSO-d₆, δ ppm): δ 58.02, 109.38, 110.01, 113.47, 115.70, 115.86, 128.82, 129.88, 129.91, 129.93, 129.94, 134.95, 142.54, 147.89, 153.08, 157.72, 159.80, 161.82, 163.06. LCMS (ESI) m/z: 354.3307 [M+1]⁺. Anal. calcd. for C₁₈H₁₂FN₃O₄S: C, 61.18; H, 3.42; N, 11.89%. Found: C, 61.17; H, 3.44; N, 11.87%.

2-(furan-2-yl)-5-(4-methylbenzylidene)-3-(pyrimidin-2-yl)thiazolidin-4-one, AZ₂: Yield 31%, solid, white, m.p. 164-166°C. IR (KBr, ν_{max}, cm⁻¹): 665 (C-S), 1226 (C-O-C), 1279 (-C-N), 1516 (-C=N-), 1726 (C=O), 2927 (C-CH₃), 3055 (C-H). ¹H NMR spectrum (500 MHz, DMSO-d₆, δ ppm): 2.41 (d, J = 0.81 Hz, 3H, C-CH₃), 6.36 (t, J = 7.52 Hz, 1H, C=CH), 6.43 – 6.48 (m, 2H), 6.93 – 7.00 (m, 1H), 7.29 – 7.35 (m, 2H), 7.44 – 7.51 (m, 3H), 7.82 (s, 1H, C=CH), 8.42 (d, J = 7.55 Hz, 2H, CH=C-CH). ¹³C NMR spectrum (125 MHz, DMSO-d₆, δ ppm): δ 23.69, 58.54, 109.90, 110.53, 113.99, 128.63, 129.34, 129.62, 131.09, 135.49, 139.41, 143.06, 148.40, 153.60, 158.24, 163.72. HRMS (ESI) m/z: 350.4136[M+1]⁺. Anal. calcd. for C₁₉H₁₅N₃O₂S: C, 65.31; H, 4.33; N, 12.03%. Found: C, 65.33; H, 4.31; N, 12.05%.

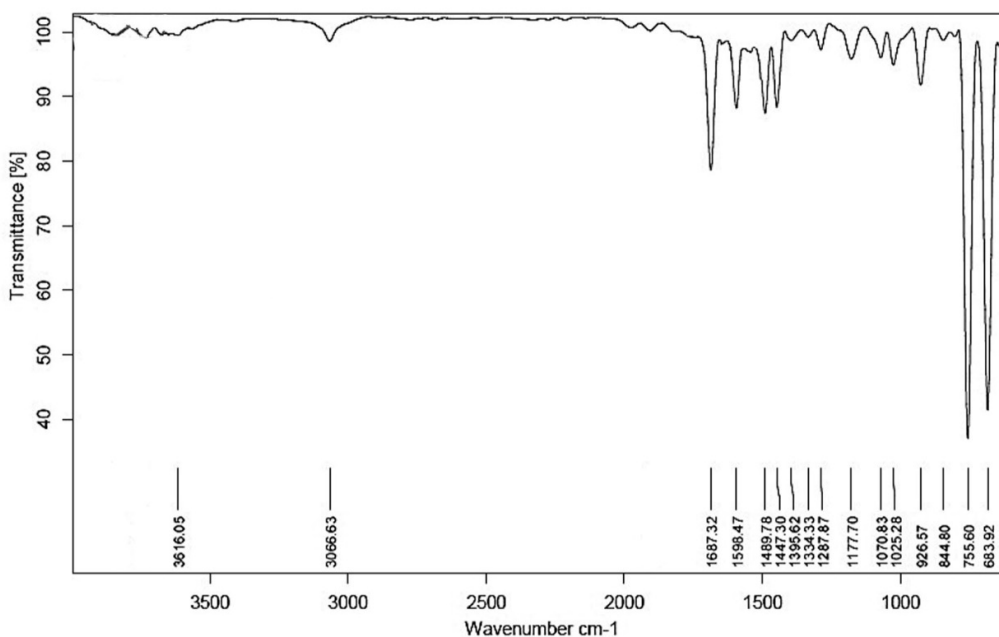


Figure 1: IR spectrum of compound AZ₁

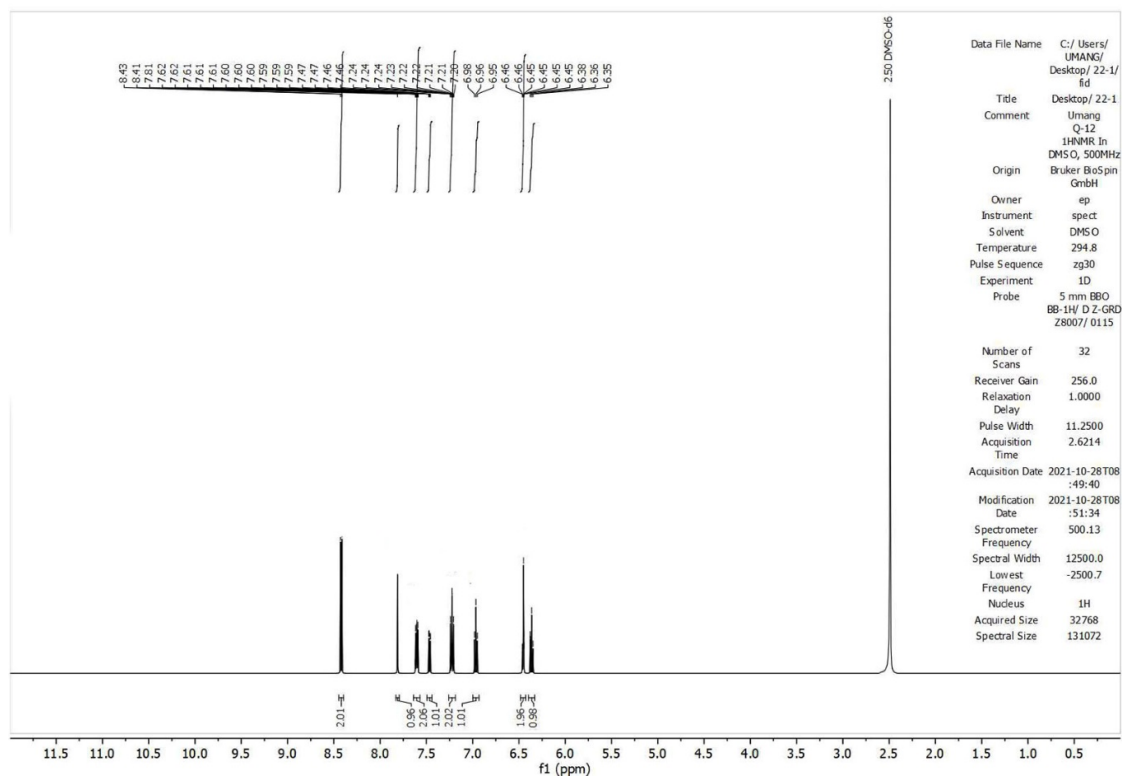


Figure 2: ^1H NMR spectrum of compound AZ₁

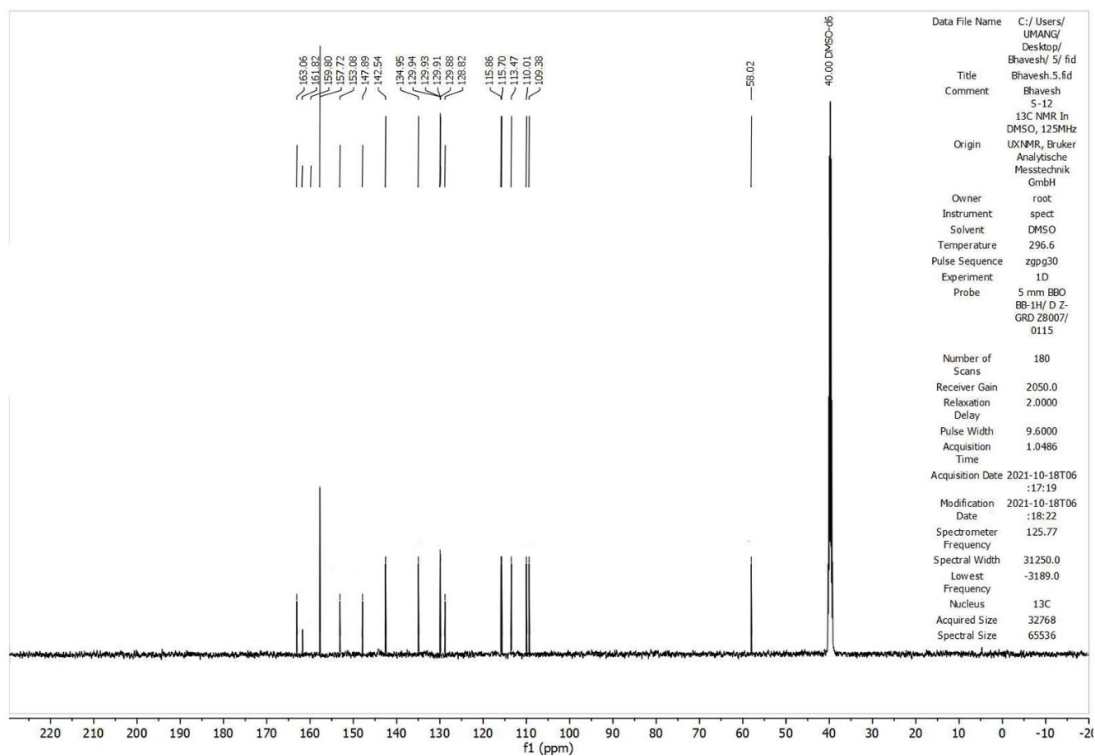


Figure 3: ^{13}C NMR spectrum of compound AZ₁

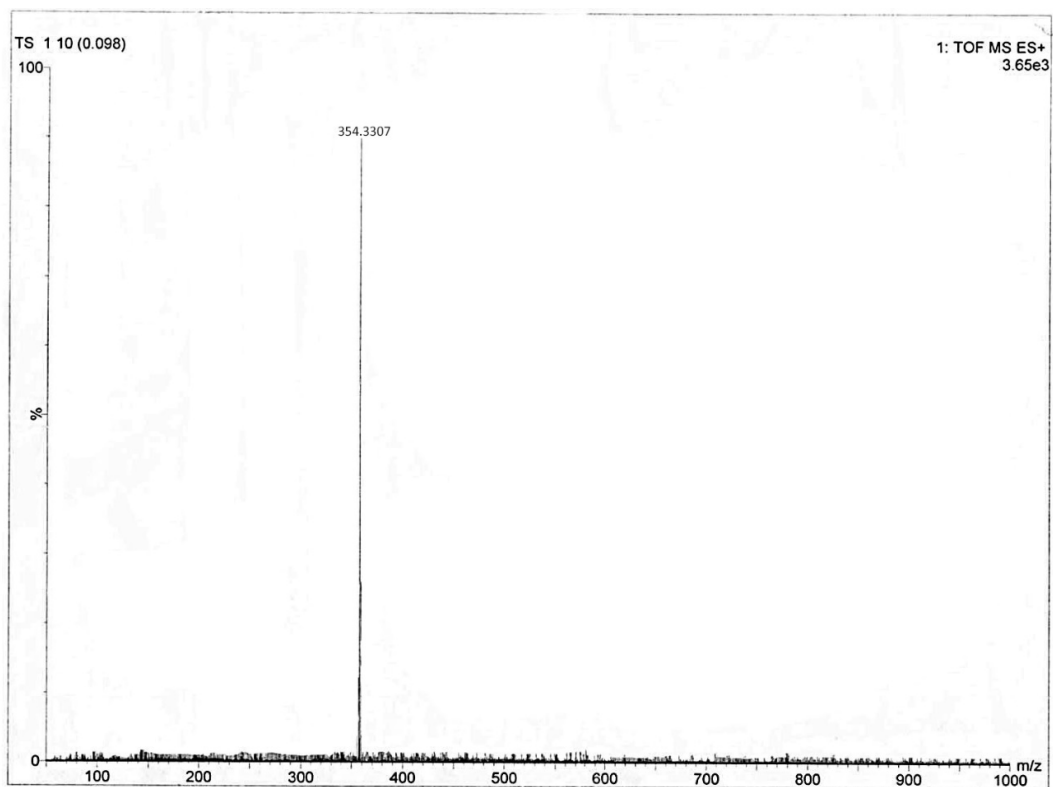


Figure 4: Mass spectrum of compound AZ₁

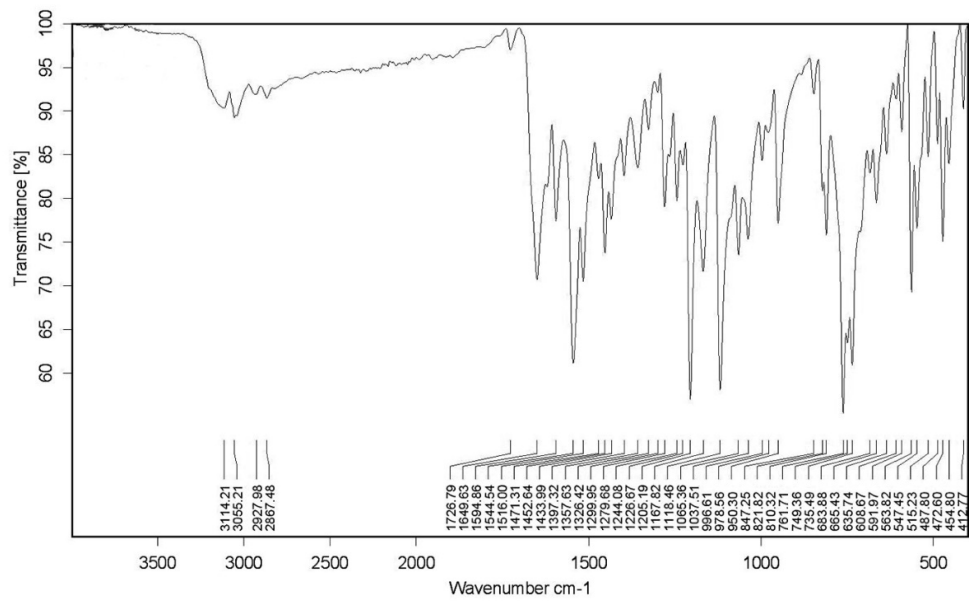


Figure 5: IR spectrum of compound AZ₂

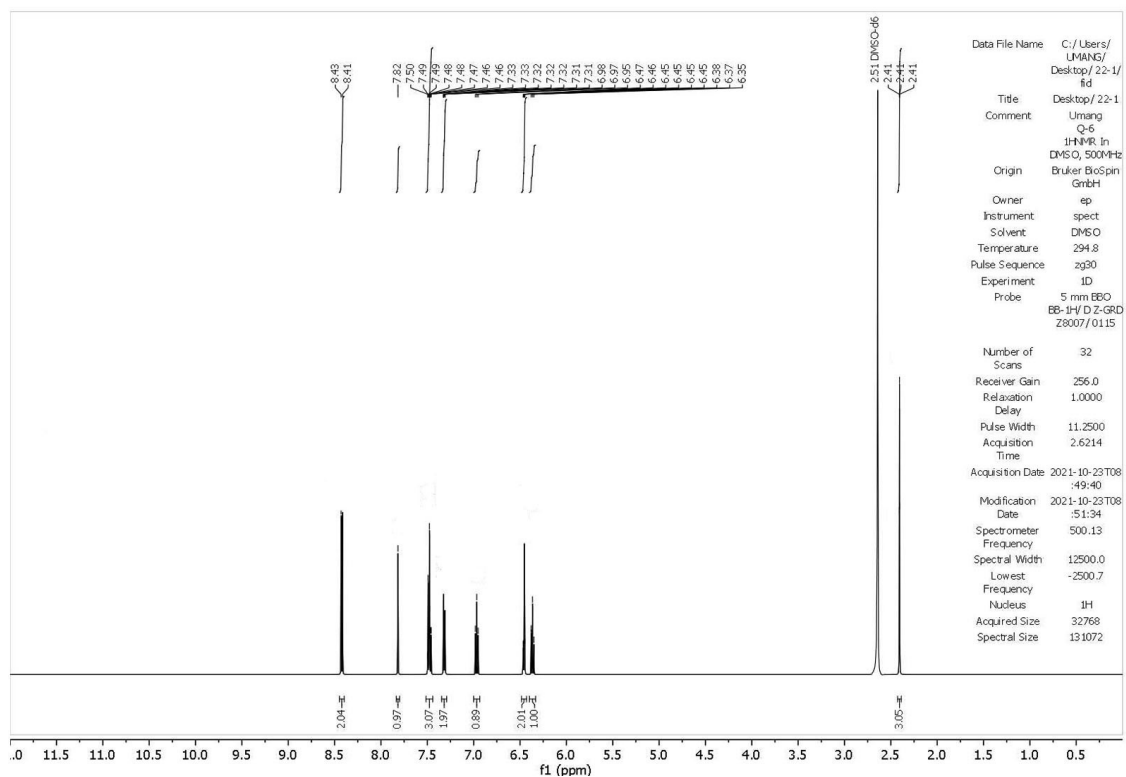


Figure 6: ^1H NMR spectrum of compound AZ₂

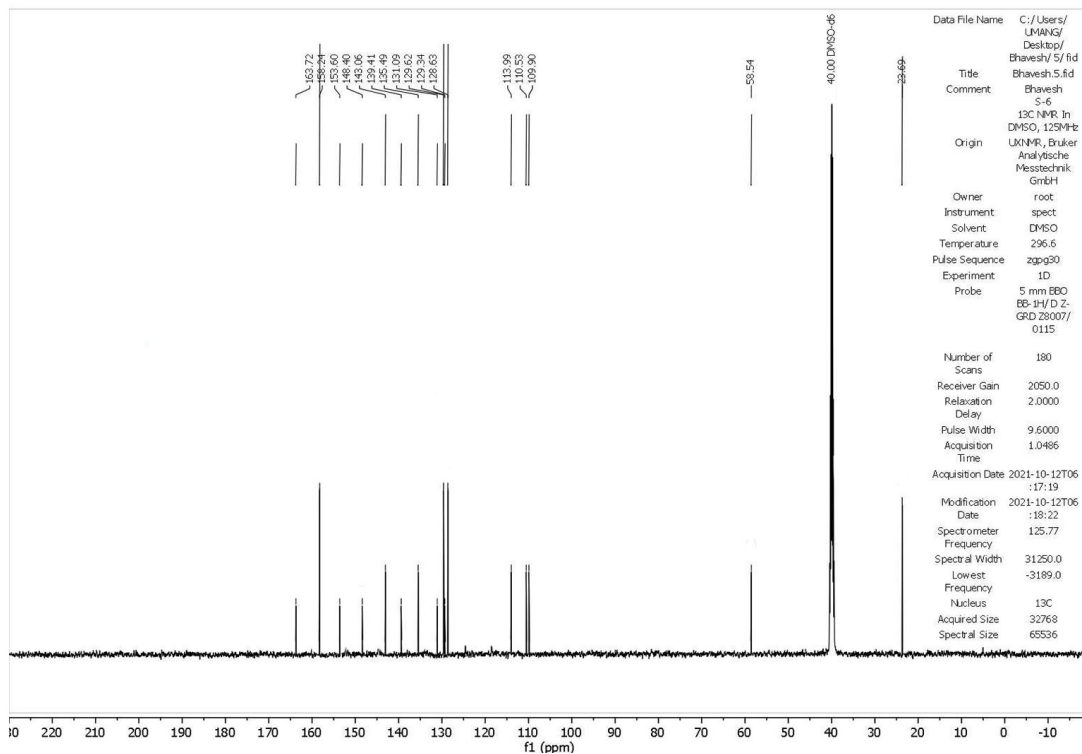


Figure 7: ^{13}C NMR spectrum of compound AZ_2

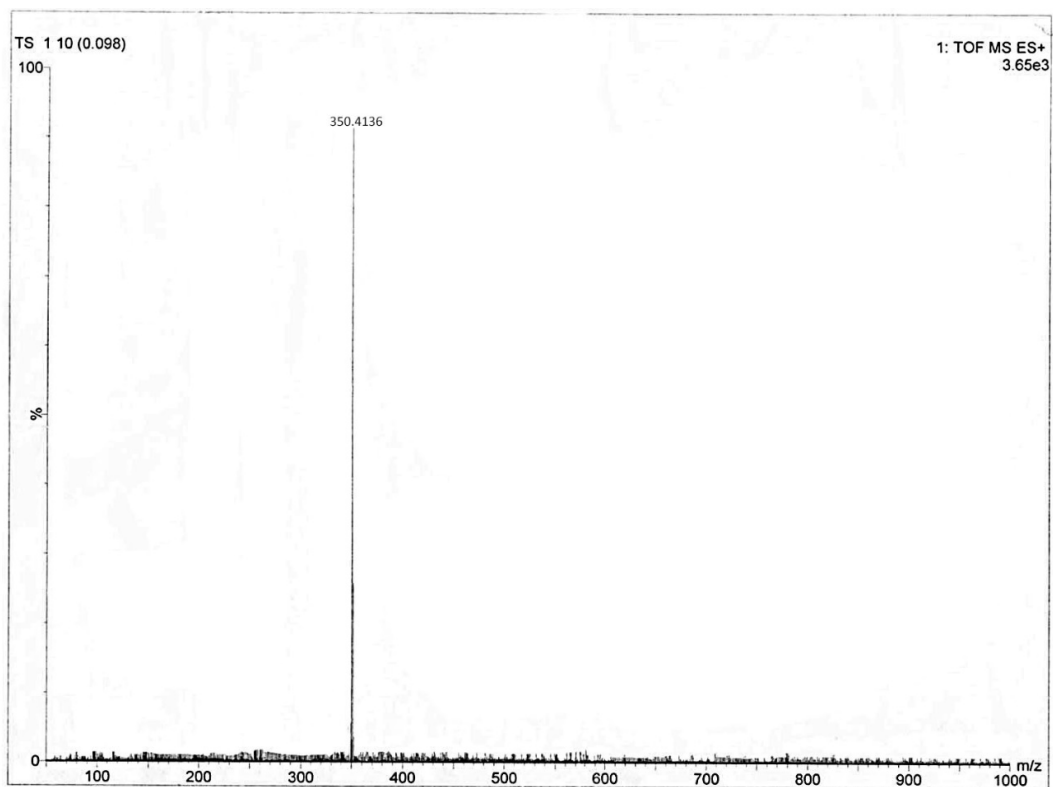


Figure 8: Mass spectrum of compound AZ₂