

## *Supplementary Information*

# Aggregation induced emissive pyrimido fused tetraphenylethene benzothiazole probe for sensitive and selective detection of Fe<sup>3+</sup> ions

Dnyaneshwar I Bhusanur<sup>a,f</sup>, Harshad A Mirgane<sup>b</sup>, Prabhat K Singh<sup>c,d</sup>, Mohammad Al Kobaisi<sup>e</sup>,  
Sheshanath V Bhosale<sup>b</sup> & Sidhanath V Bhosale<sup>\*a,f</sup>

<sup>a</sup> Polymers and Functional Materials Division, CSIR-Indian Institute of Chemical Technology, Hyderabad 500 007, Telangana, India

<sup>b</sup> Department of Chemistry, School of Chemical Sciences, Central University of Karnataka, Kadaganchi, Kalaburagi 585 367, Karnataka, India

<sup>c</sup> Radiation and Photochemistry Division, Bhabha Atomic Research Centre, Mumbai 400 085, India

<sup>d</sup> Homi Bhabha National Institute, Training School Complex, Anushaktinagar, Mumbai 400 094, India

<sup>e</sup> School of Science, RMIT University, GPO Box 2476, Melbourne, VIC, 3001, Australia

<sup>f</sup> Academy of Scientific and Innovative Research (AcSIR), CSIR-HRDC Campus, Postal Staff College Area

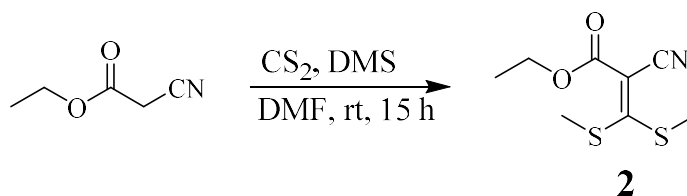
Sector 19, Kamla Nehru Nagar, Ghaziabad 201 002, Uttar Pradesh, India

E-mail: bhosale@iict.res.in

*Received 22 May 2024; accepted (revised) 30 August 2024*

## Experimental Section

### Synthesis of ethyl 2-cyano-3,3-bis(methylthio)acrylate (**2**):



#### Scheme S1. Synthesis of ethyl 2-cyano-3,3-bis(methylthio)acrylate (**2**):

To a mixture of potassium hydroxide (0.424 g, 10.60 mmol) in 15 ml of dry DMF the reaction mixture was stirred at r.t. for 20 min. Then ethyl cyanoacetate (1 g, 8.84 mmol) was added dropwise at r.t. The resulting mixture was stirred at r.t for 1 h and carbon disulfide (0.8077 g, 10.60 mmol) was added at 0 °C. The mixture stirred at 0 °C for 1.5 h and dimethyl sulfate (2.23 g, 17.68 mmol) was added slowly at 0 °C. Then the resulting reaction mixture was stirred at r.t for 12 h and after completion of the reaction was monitored by TLC plate. The above reaction mixture was poured in to ice water with constant stirring and compound **2** was precipitated out. Further washed with water and dried to afford a yellow solid (**2**, 60%). Then it was used without further purification. FT-IR (cm<sup>-1</sup> KBr) 2210, 1661, 1445, 1316, 1210, 935; <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>) δ ppm: 4.3 (q, 2H, *J* = 7.2 Hz), 2.75 (s, 3H), 2.6 (s, 3H), 1.37 (t, 3H, *J* = 7.09 Hz); <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>) δ 180.8, 162.4, 116.1, 98.8, 61.8, 20.9, 18.9, 14.2; ESI-mass *m/z*: calcd 217.30 [M]<sup>+</sup>, found 218.00 [M+H]<sup>+</sup>; HRMS *m/z*: [M+H]<sup>+</sup> calcd. for C<sub>8</sub>H<sub>12</sub>NO<sub>2</sub>S<sub>2</sub> 218.03040; found 218.03134 [M+H]<sup>+</sup>.

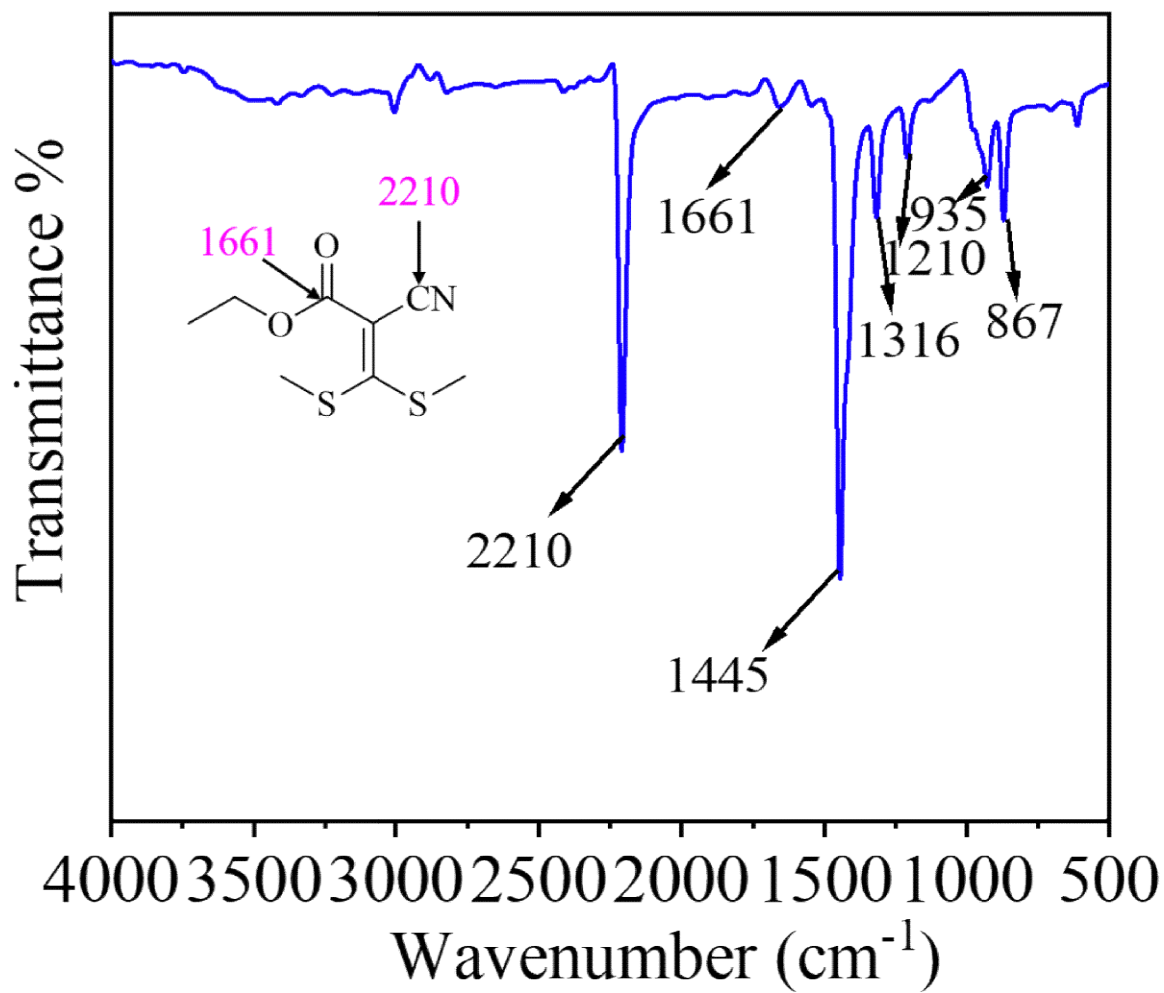


Figure S1. FT-IR spectra of compound 2

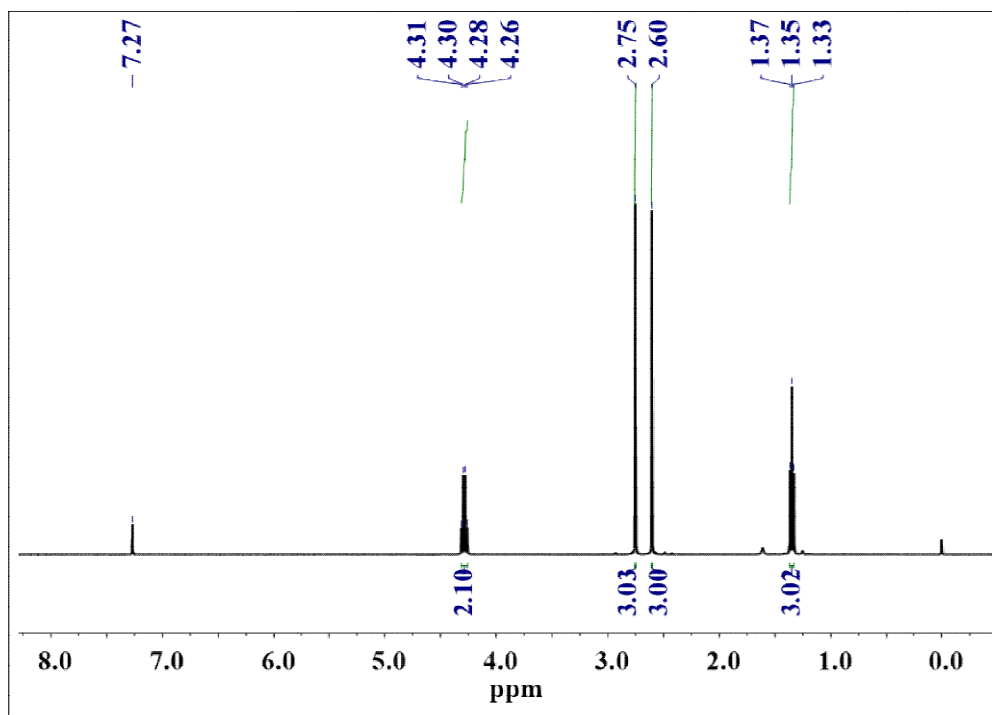


Figure S2. <sup>1</sup>H NMR spectra of compound 2.

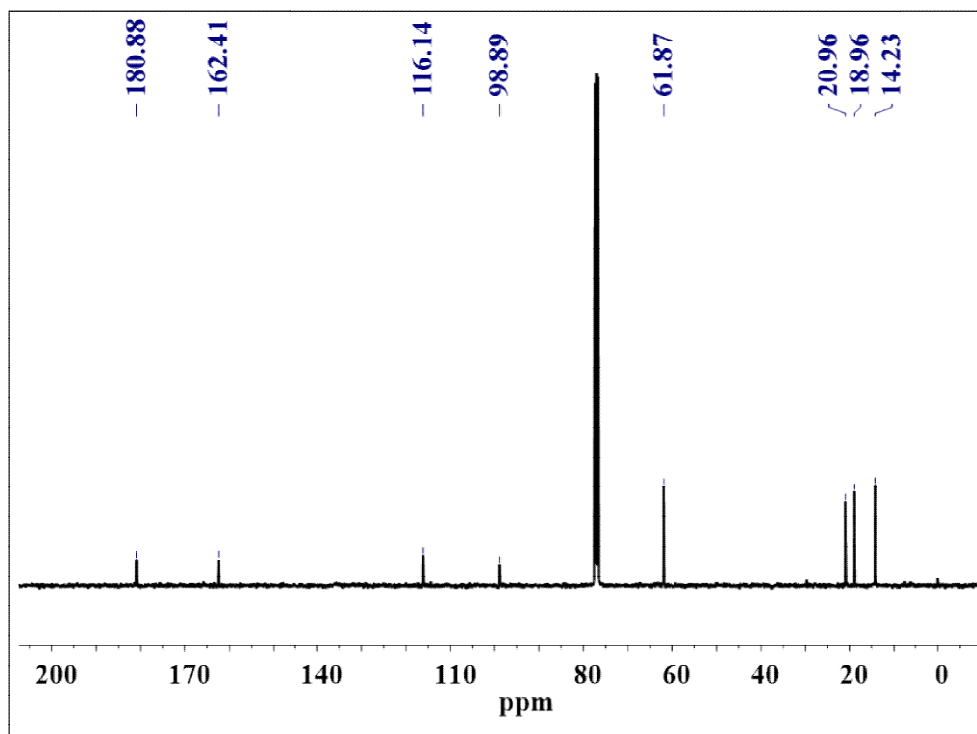
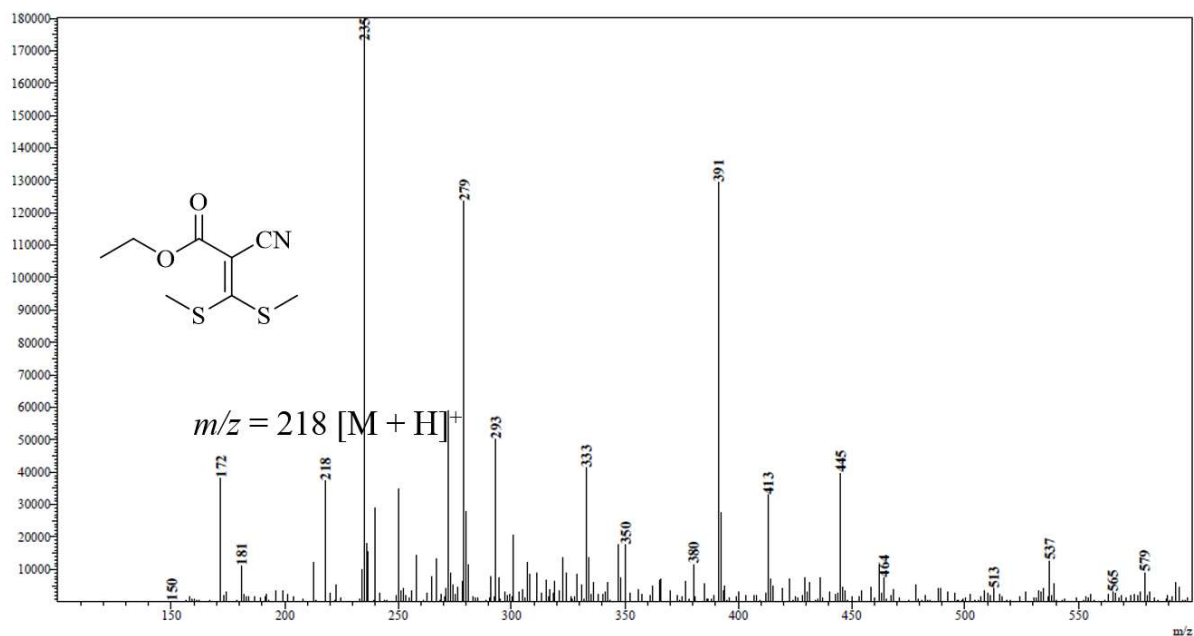
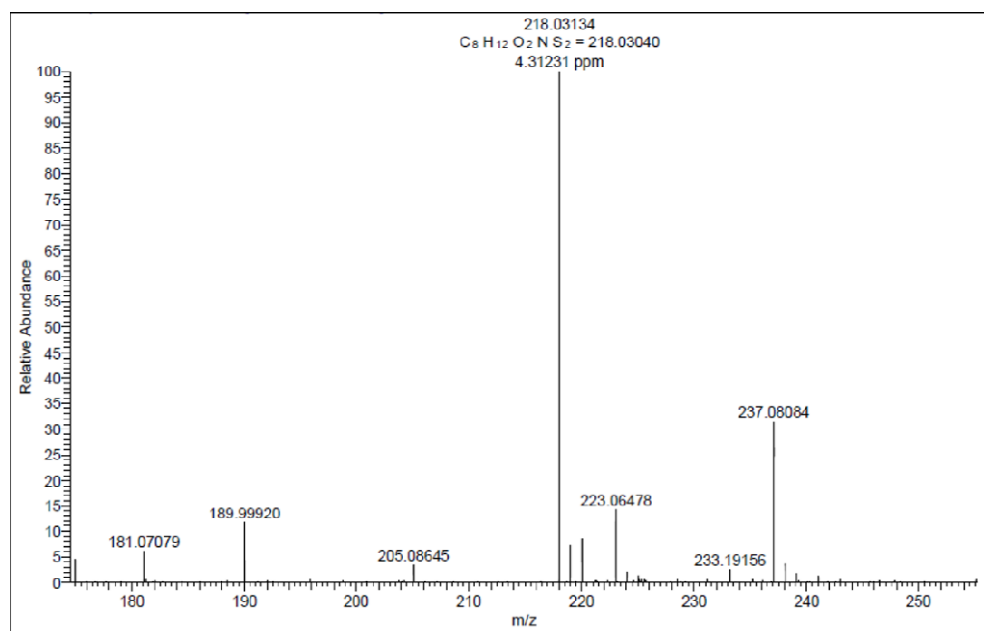


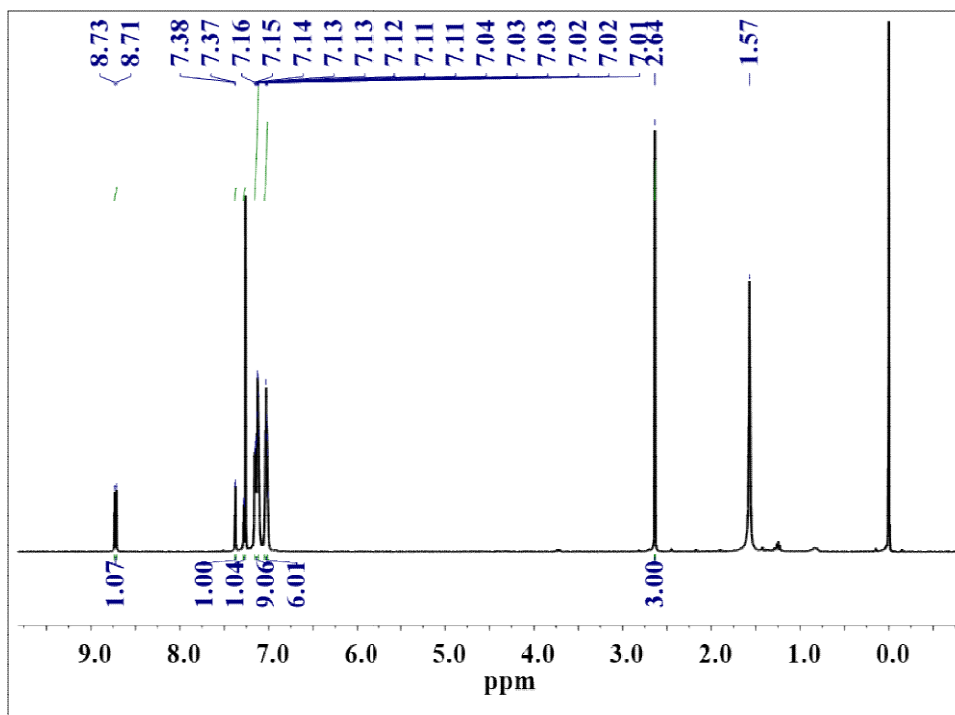
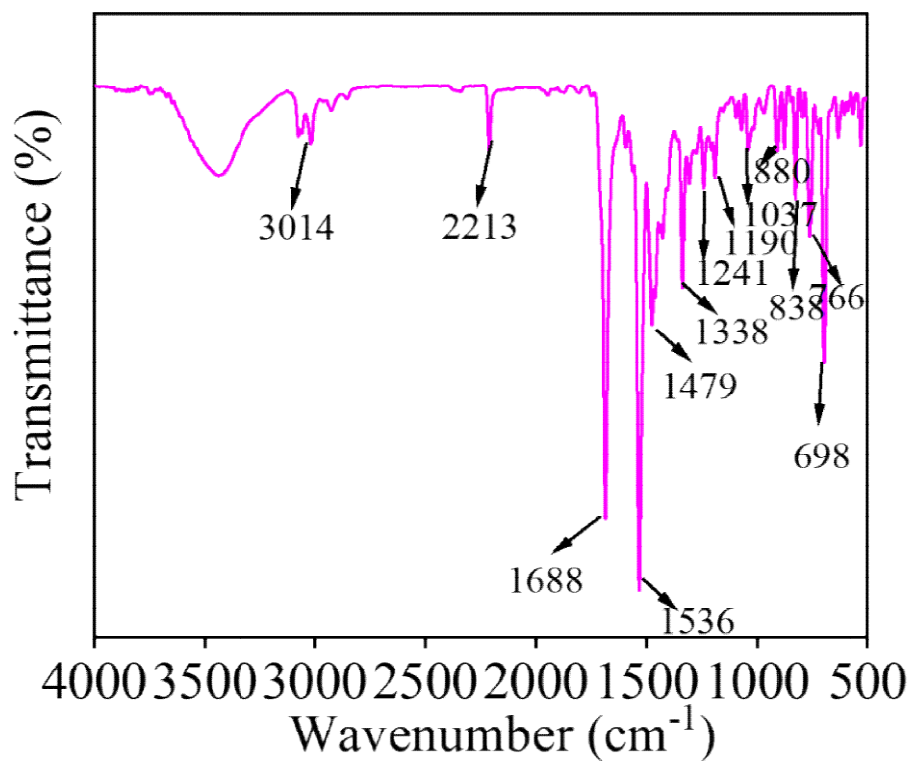
Figure S3. <sup>13</sup>C NMR spectra of compound 2.



**Figure S4.** Mass spectrum of compound 2.

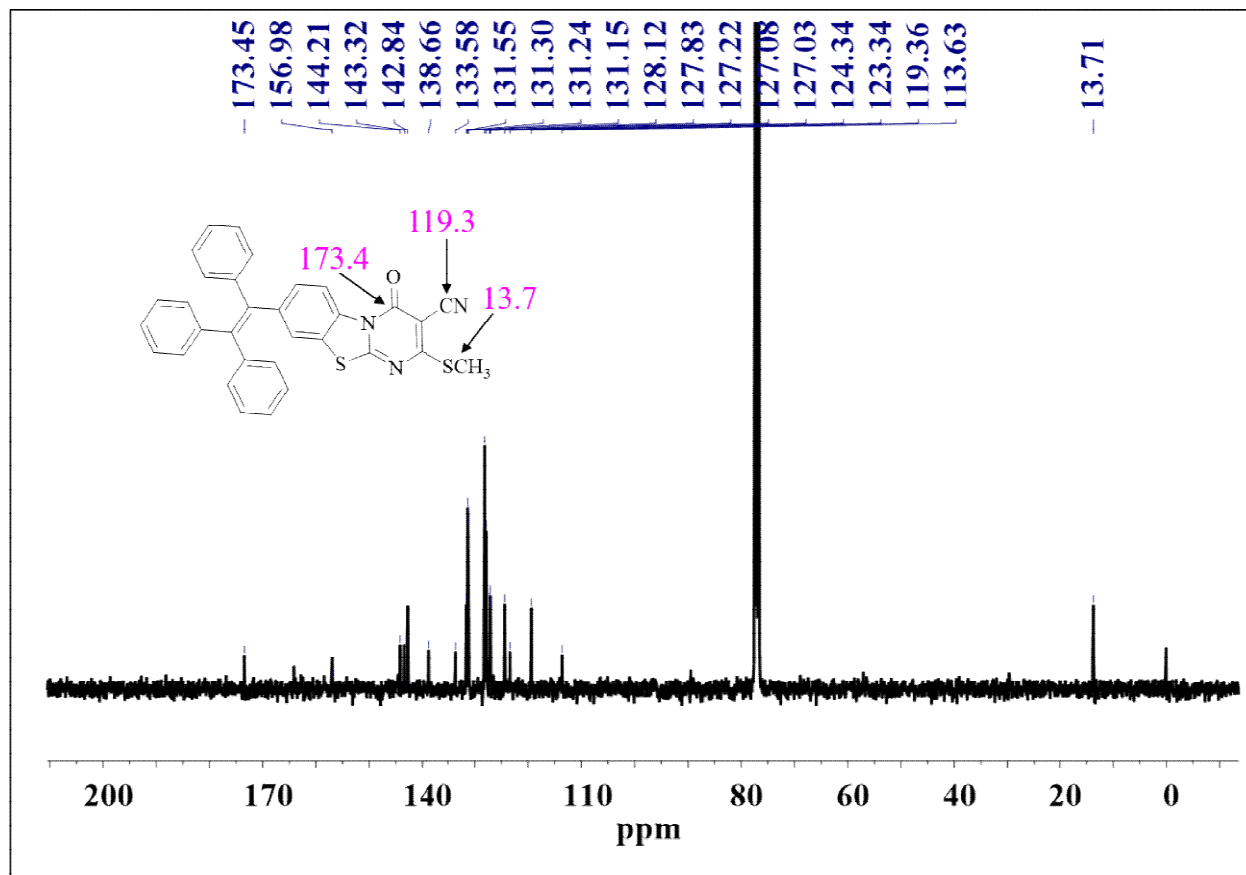


**Figure S5.** HRMS spectrum of compound 2.

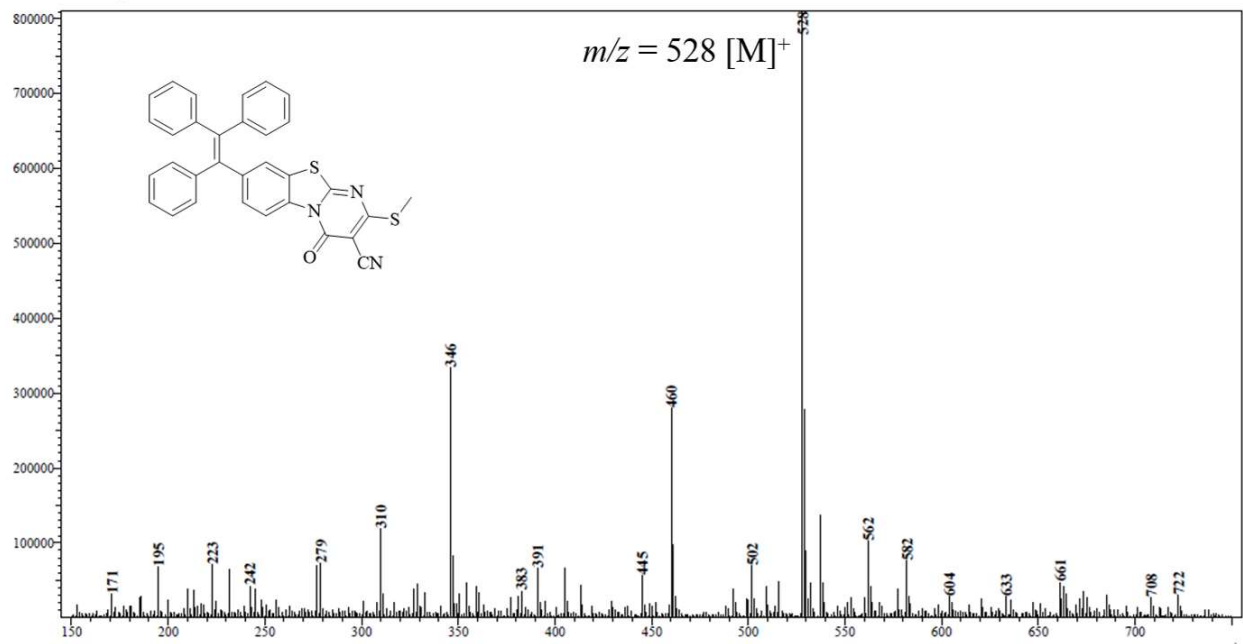


**Figure S6.** FT-IR spectra of compound TPE-1

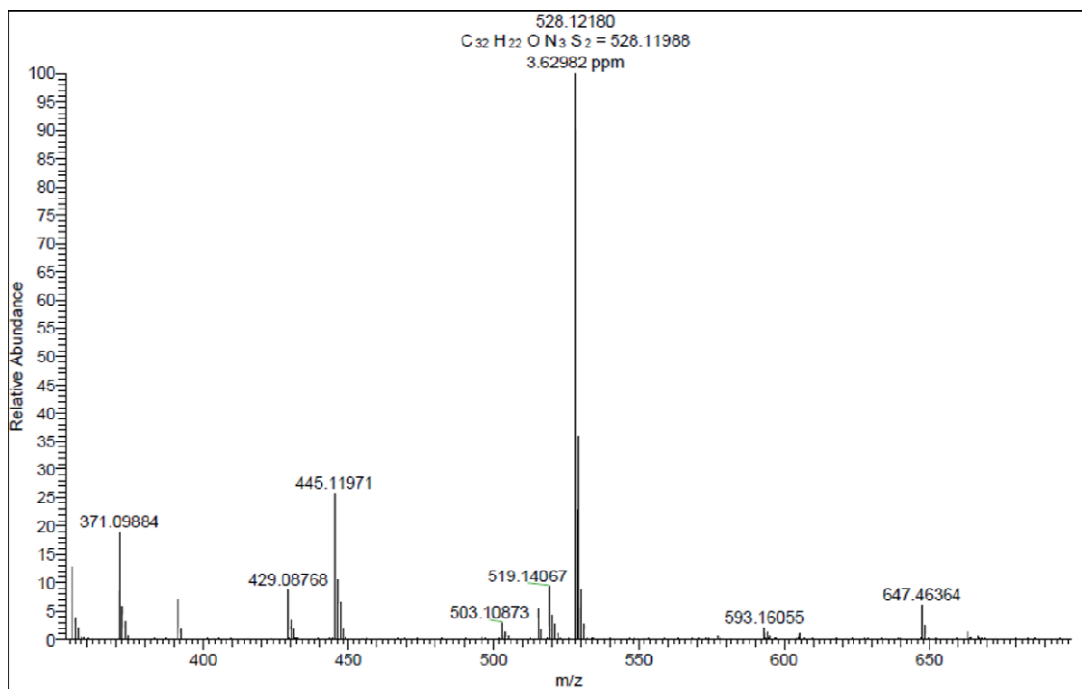
**Figure S7.**  $^1\text{H}$  NMR spectra of compound TPE-1



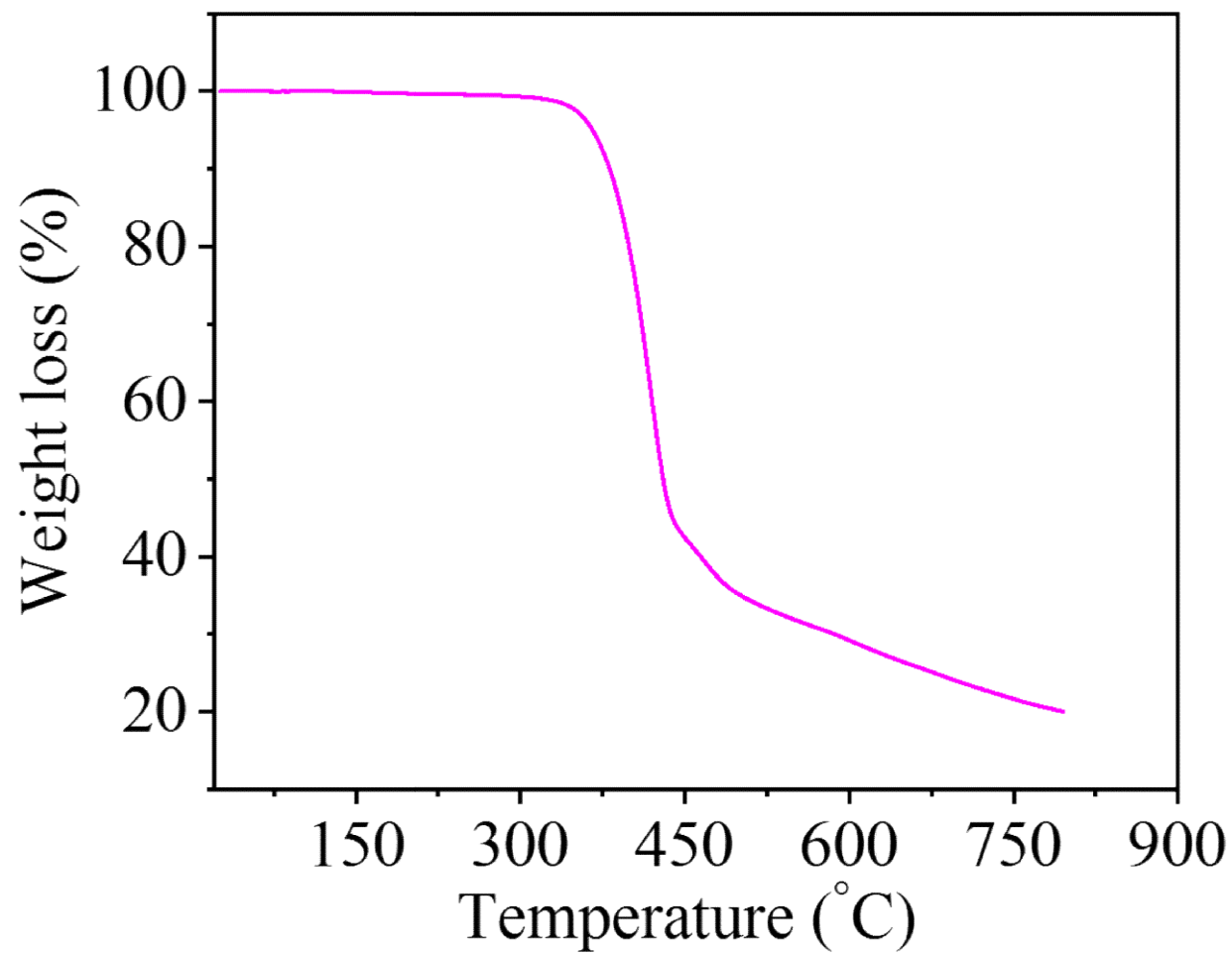
**Figure S8.**  $^{13}\text{C}$  NMR spectra of compound TPE-1



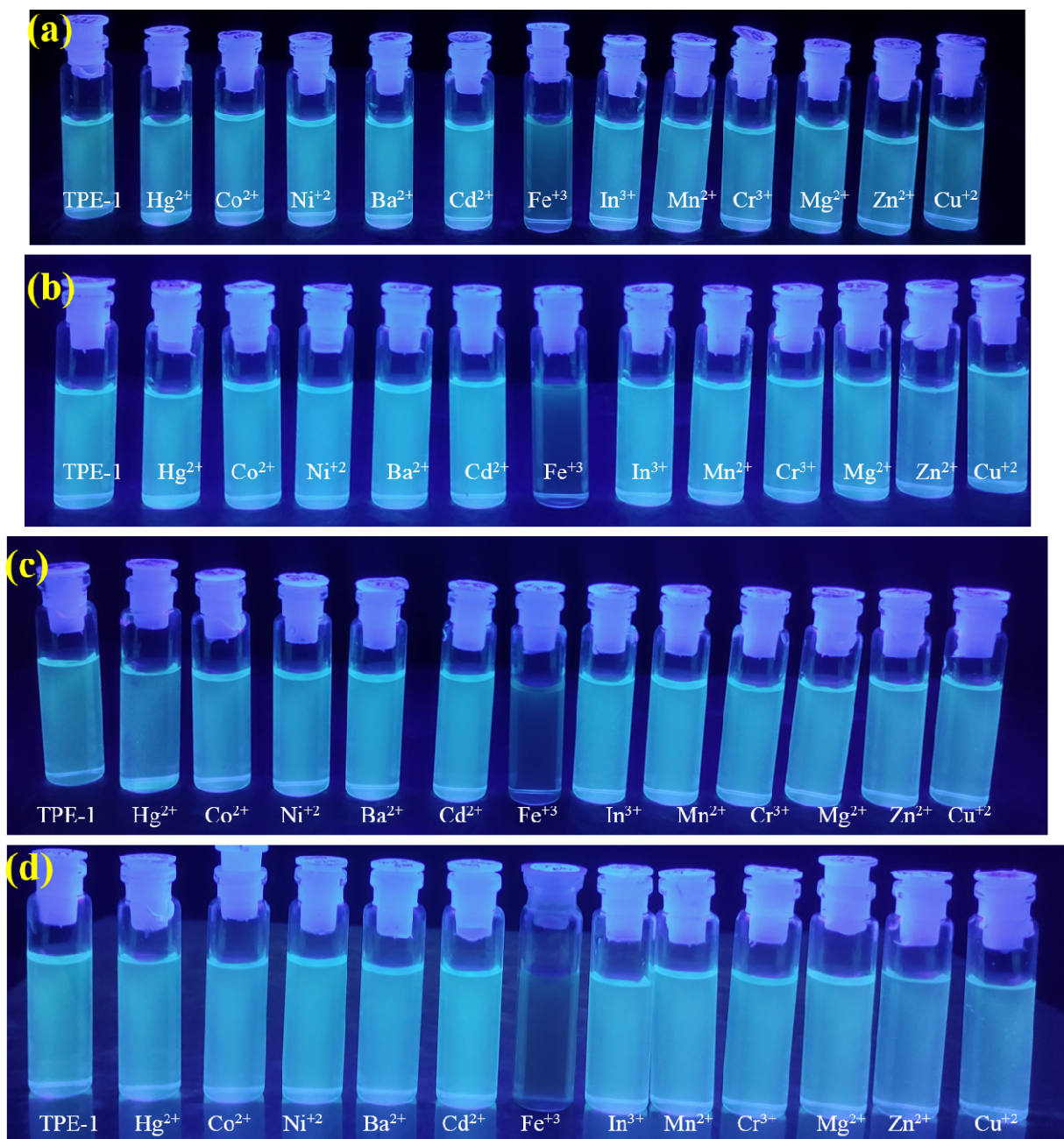
**Figure S9.** Mass spectrum of TPE-1.



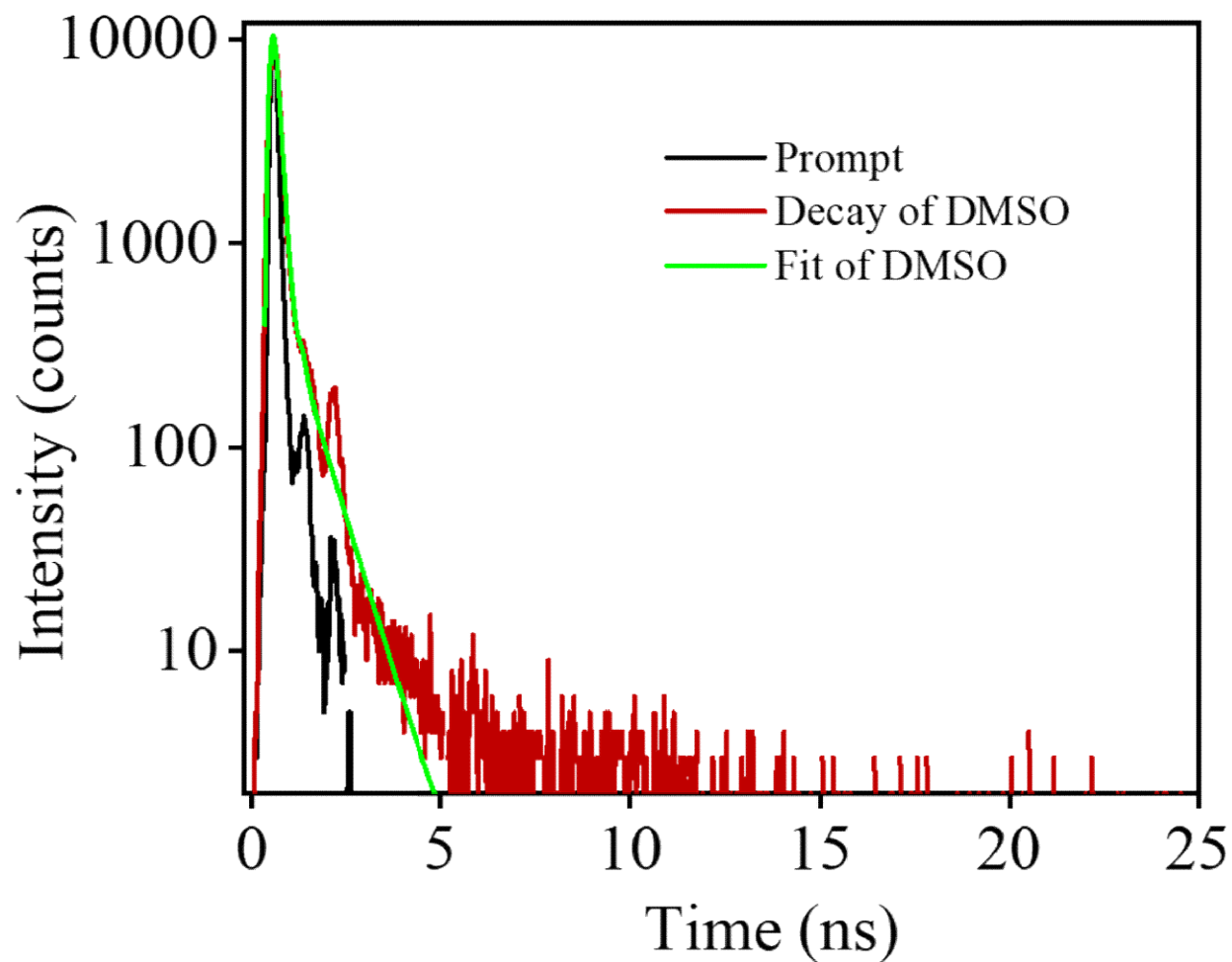
**Figure S10.** HRMS spectrum of compound TPE-1.



**Figure S11.** Thermogravimetric analysis (TGA) of TPE-1.



**Figure S12.** Photographic images of TPE-1 with (a) 100 equiv.; (b) 200 equiv., (c) 300 equiv. and (d) 500 equiv. of various metals ions in DMSO: H<sub>2</sub>O (10:90).



**Figure S13.** TCSPC measurements of TPE-1 ( $1 \times 10^{-5}$  M) in DMSO.